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SECURITY CLASSIFICATION OF THIS PAGE

## REPORT DOCUMENTATION PAGE

Form Approved  
OMB No 0704-0188

AD-A220 995

DTIC  
ELECTE  
APR 30 1990

1b RESTRICTIVE MARKINGS

NA

3 DISTRIBUTION/AVAILABILITY OF REPORT

UNLIMITED

4 PERFORMING ORGANIZATION REPORT NUMBER(S)

NA

5 MONITORING ORGANIZATION REPORT NUMBER(S)

NA

6a NAME OF PERFORMING ORGANIZATION

UCLA

6b OFFICE SYMBOL  
(If applicable)

NA

7a NAME OF MONITORING ORGANIZATION

OFFICE OF NAVAL RESEARCH

6c ADDRESS (City, State, and ZIP Code)

405 Hilgard Avenue  
Los Angeles, California 90024

7b ADDRESS (City, State, and ZIP Code)

800 N. Quincy Street  
Arlington, Virginia 22217-50008a NAME OF FUNDING/SPONSORING  
ORGANIZATION

OFFICE OF NAVAL RESEARCH

8b OFFICE SYMBOL  
(If applicable)

9 PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER

N00014-86-K-0525

8c ADDRESS (City, State, and ZIP Code)

800 N. Quincy Street  
Arlington, Virginia 22217-5000

10 SOURCE OF FUNDING NUMBERS

PROGRAM  
ELEMENT NOPROJECT  
NOTASK  
NOWORK UNIT  
ACCESSION NO

11 TITLE (Include Security Classification)

Structure and Design of Multipotent Peptide Microbicides

12 PERSONAL AUTHOR(S)

Michael E. Selsted

13a TYPE OF REPORT

Final

13b TIME COVERED

FROM 11/86 TO 11/89

14 DATE OF REPORT (Year, Month, Day)

1990, April 19

15 PAGE COUNT

16 SUPPLEMENTARY NOTATION

17 COSATI CODES

FIELD

GROUP

SUB-GROUP

06

03

18 SUBJECT TERMS (Continue on reverse if necessary and identify by block number)

Peptides, Structure-Activity Relationships,  
Synthesis, Microbicides, antimicrobial peptides (AT)

19 ABSTRACT (Continue on reverse if necessary and identify by block number)

Defensin peptides, naturally occurring neutrophil antibiotics, were purified and characterized functionally and structurally. Functional differences in microbicidal potency and spectrum were correlated with structural features resolved by NMR and crystallographic methods. Protocols were established for synthesis and disulfide formation of natural defensins and synthetic congeners; we are currently investigating the latter by structural and functional characterization.

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20 DISTRIBUTION/AVAILABILITY OF ABSTRACT

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21 ABSTRACT SECURITY CLASSIFICATION

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22b TELEPHONE (Include Area Code)

202-696-4760

22c OFFICE SYMBOL

ONR

FINAL TECHNICAL REPORT  
OFFICE OF NAVAL RESEARCH  
CONTRACT N00014-86-K-0525  
11-1-86 TO 10-31-89

Defensins, homologous antimicrobial peptides isolated from neutrophils of several mammalian species, were employed in structure-activity studies aimed at the design of novel antimicrobial peptides. As defensins are small (29-34 residues) and constrained by three intramolecular disulfides, they are all thought to possess the same overall fold. Therefore, these peptides are ideally suited to SAR studies. Further, the naturally-occurring functional (microbicidal potency and spectrum) variation among members of this family provides a "head-start" in devising a scheme for dissecting structure and function.

During the term of the contract, a number of new defensins were discovered by us and others. The family of defensin peptides now known is shown in Fig. 1.

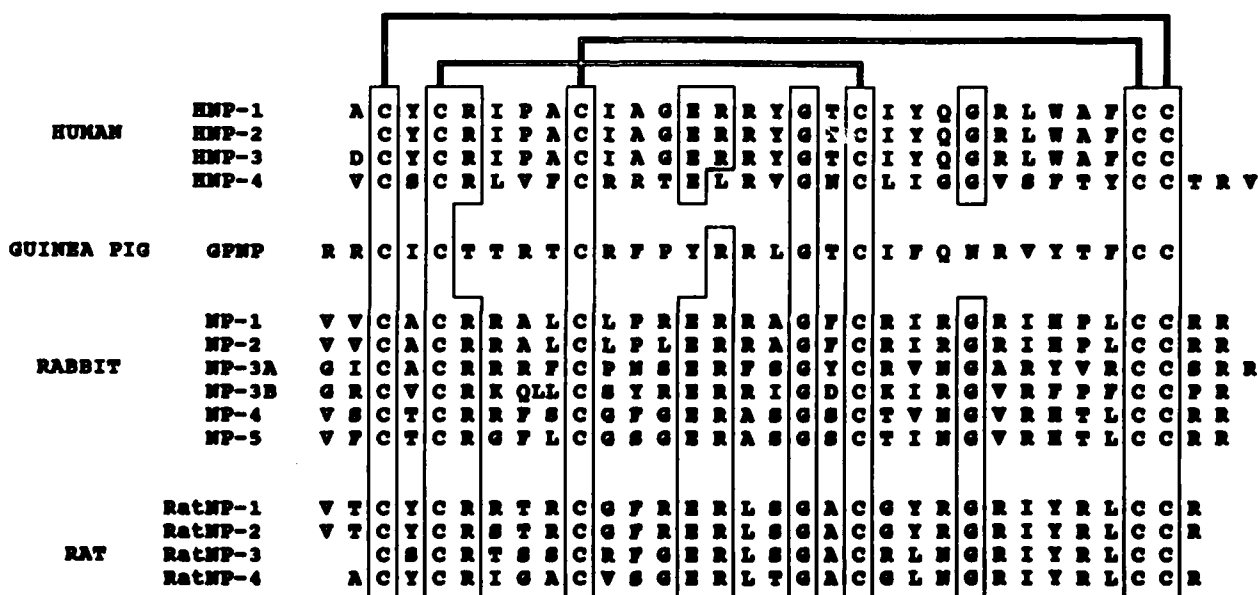
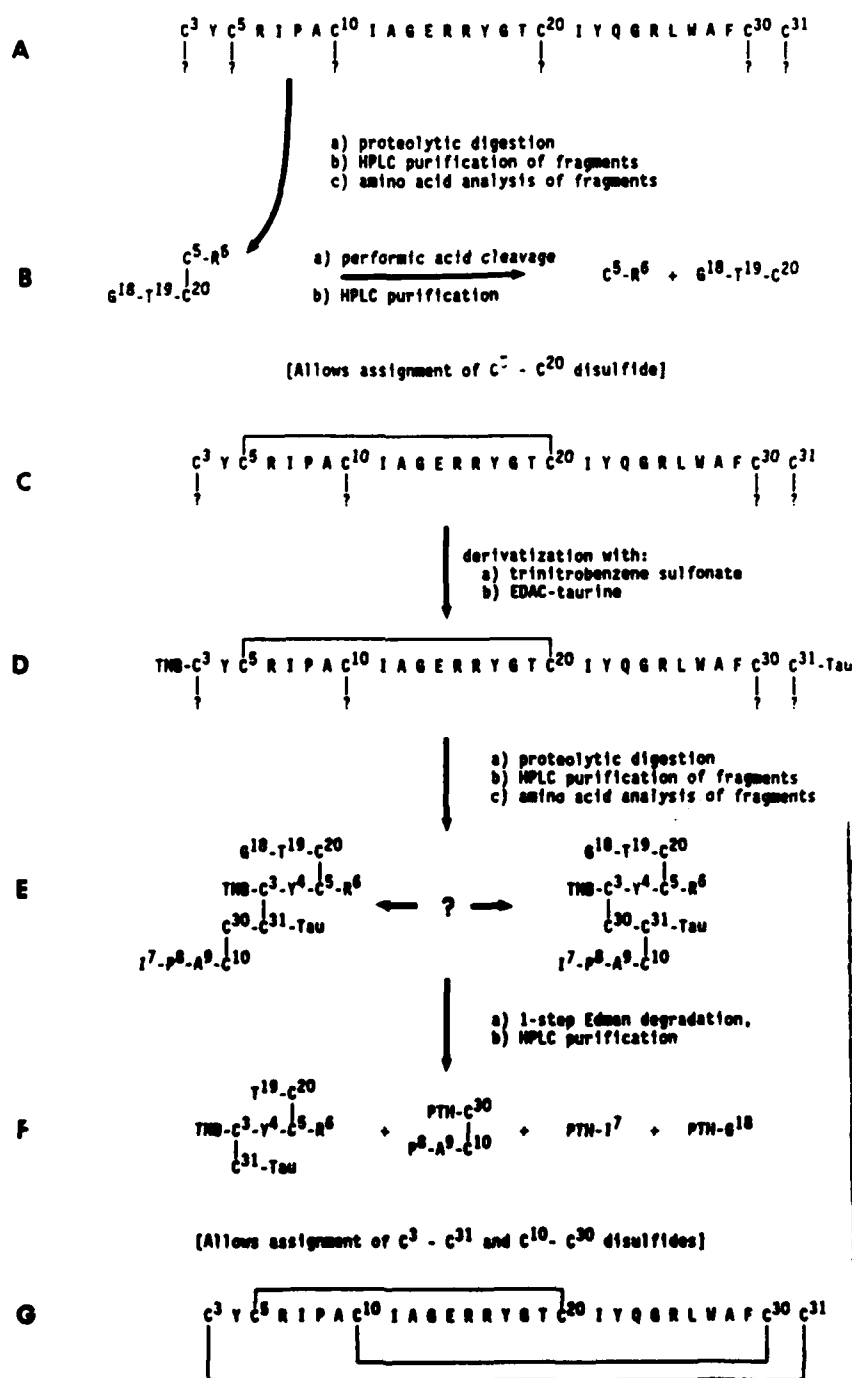


Figure 1. Amino acid sequences and disulfide connectivities of defensins.

The disulfide connectivities were determined using a novel biochemical approach, revealing that the disulfide motif in defensins is unique. Note that in HNP-2, a disulfide connects the amino and carboxyl terminal residues, a feature not known to occur in any other protein. (Fig. 2).



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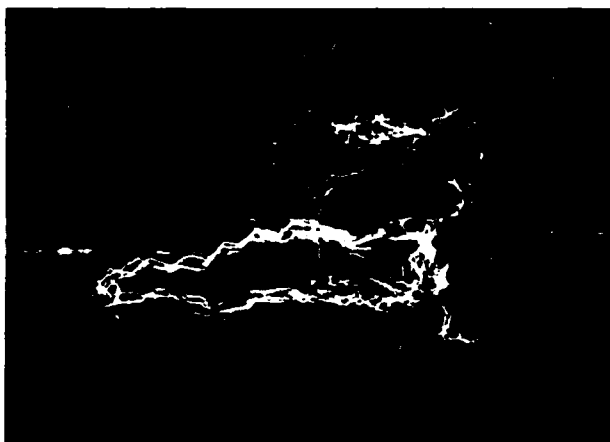
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Fig. 2. Determination of the disulfide motif in defensin HNP-2.

Structural studies of defensins were carried out in solution and in crystals. Two dimensional NMR revealed that HNP-1, NP-2, and NP-5 were very similar in their overall fold, and that each forms a three stranded antiparallel sheet (Fig. 3).

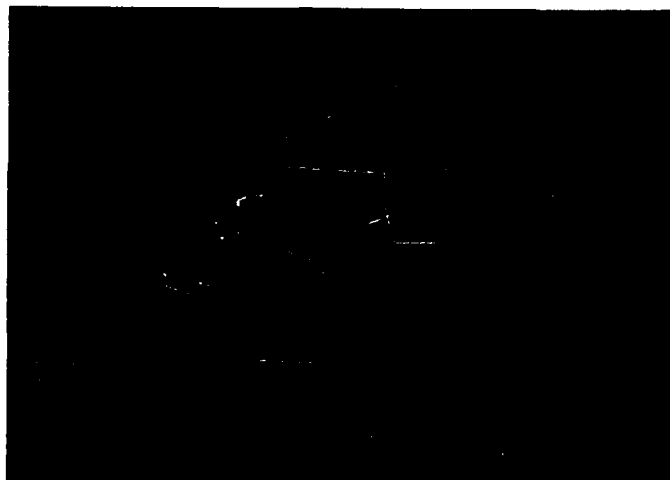


**Figure 3** Comparison of the backbone conformation of HNP-1, NP-2 and NP-5. The main chains of HNP-1 (red), NP-2 (yellow) and NP-5 (blue) were superimposed by calculating the best fit for C, N, O and C $\alpha$  atoms for residues 16-31 in each peptide.

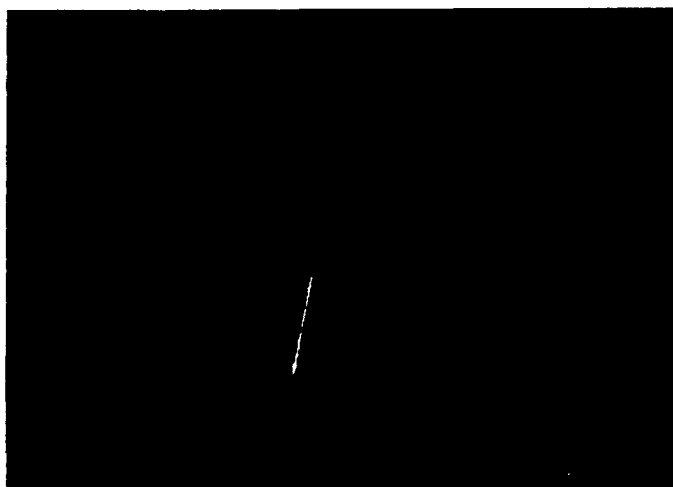
Of the several defensin peptides which have been crystallized, HNP-3 was found to be most useful for diffraction studies. Platinum salt derivatives of HNP-3 were used to solve the structure of this peptide at 1.9 Å ( $R_{\text{factor}} = 0.19$ ; Fig. 4). The crystal structure of HNP-3 revealed a dimeric structure of this defensin, and confirmed both the disulfide assignments and the solution conformation. The dimer of HNP-3 is amphiphilic, as there is a large patch of leucine and isoleucine residues concentrated at one pole, while the N and C termini of both chains are positioned at the opposite pole. The amphiphilic topology is thought to correlate with the ability of defensins to permeabilize lipid bilayers.

One of the major goals of the project was to utilize solid phase synthesis for producing defensin analogs. We utilized both BOC and Fmoc chemistries to synthesize NP-2, HNP-1, and GPNP. Although refolding and oxidation of each peptide required very different conditions, we were able to derive conditions for each peptide which resulted in quantitative, correct disulfide formation. The first set of synthetic congeners has been produced, and we are currently evaluating the antimicrobial activities of these analogs.

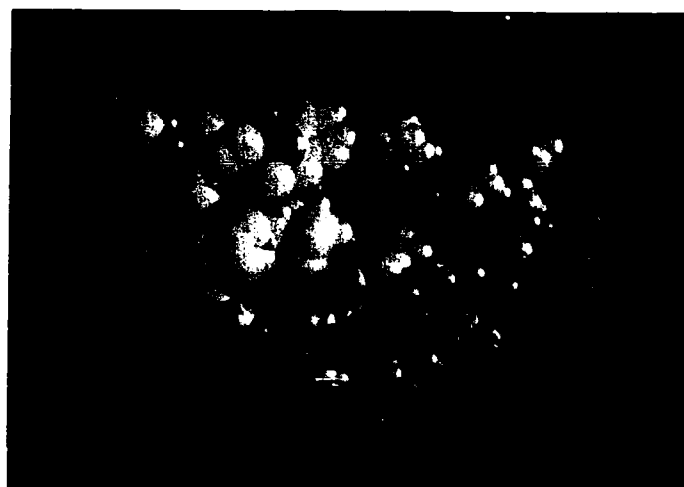
A



B



C



**Figure 4.** The structure of human defensin HNP-3 at 1.9 Å. Two platinum derivatives were used to derive the structure (25). A. The main chain (green), the cystine disulfides (yellow), and the arginine side chains are depicted. B & C. All atoms of the dimer are shown in skeleton or space filling projection. The chain termini are oriented toward the top of the page, and the  $\beta$  hairpin at the bottom. The arrows indicate the vector of the hydrophobic moment. Atomic color code for B: carbon - yellow; amide N - pink; charged N - red; uncharged O - light blue; charged O - dark blue; S - charcoal.

PUBLICATIONS ACKNOWLEDGING ONR  
CONTRACT N00014-86-K-0525  
11-1-86 to 10-31-89

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